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Aryne approach towards 2,3-difluoro-10-(1*H*-1,2,3-triazol-1-yl)pyrido[1,2-*a*]indoles

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Reaction between 3-(2-pyridyl)-1,2,4-triazines and *in situ* generated 4,5-difluorobenzyne in toluene affords 2,3-difluoro-10-(1*H*-1,2,3-triazol-1-yl)pyrido[1,2-a]indoles. The structure of one representative compound was confirmed by X-ray diffraction analysis.

Pyrido[1,2-*a*]indoles are of practical interest due to their cytostatic,¹ antiviral² and antitumor³ activities. Introduction of fluorine atoms into poly(hetero)cyclic systems in most cases improves their photophysical properties,⁴ biological activity⁵ and electrochemical properties.⁶ For now fluorinated pyrido[1,2-*a*]indoles are known scarcely,⁷ while substituted 2,3-difluoropyrido[1,2-*a*]indoles are not reported. As for 1,2,3-triazoles, they are of interest⁸ as common pharmacophores⁹ as well as components of photoluminescent sensors for various analytes,¹⁰ including explosives.¹¹

The methods for the synthesis of pyrido[1,2-a]indoles are mostly limited by availability of the starting materials and/or the application of harsh conditions.¹² As for 1,2,3-triazole-substituted pyrido[1,2-a]indoles, recently the convenient one-pot approach *via* the benzyne-mediated rearrangement of 3-(2-pyridyl)-1,2,4triazines was reported by our group.¹³ In this paper the convenient one-pot approach to 10-substituted 2,3-difluropyrido[1,2-a]indoles by reaction between 3-(2-pyridyl)-1,2,4-triazines and 4,5-difluorobenzyne is reported.

During the last decade, aryne chemistry is experiencing a renaissance, due to the development of new methods for the smooth and effective aryne generation.¹⁴ In particular, 4,5-di-fluorobenzyne can be generated from 4,5-difluoro-2-trimethyl-silylphenyl trifluoromethanesulfonate¹⁵ by the action of fluoride anion or from 1,2-dibromo-4,5-difluorobenzene¹⁶ by the action of *n*-butyllithium. On the other hand, 3,4-difluoroanthranilic acid, which can be easily obtained by 3,4-difluorobenzyne source.¹⁸ In this work, 3,4-difluoroanthranilic acid–derived 4,5-difluorobenzyne is used to prepare 2,3-difluor-10-(1*H*-1,2,3-triazol-1-yl)pyrido[1,2-*a*]indoles in one step in good yields.

For the synthesis of 3,4-difluoroanthranilic acid 1, the modified method based on 5,6-difluoroisatine 2^{19} was used (Scheme 1).

1,2,4-Triazines **3** were synthesized according to reported procedures²⁰ by the reaction of isonitrosoacetophenone hydrazones with pyridine-2-carbaldehyde or by the reaction between α -diketones and amidrazones. The fluorination of the aryl substituent of the triazine core as well as the aryl precursor of the 4,5-difluorobenzyne did not influence the reaction pathway giving 2,3-difluoro-10-(1*H*-1,2,3-triazol-1-yl)pyrido[1,2-*a*]indoles **4** in moderate yields (see Scheme 1).[†] No isoquinolines **5** were observed in a



Scheme 1 Reagents and conditions: i, NaOH, H_2O_2 , 55 °C, 2 h, then HCl, 15–20 °C; ii, toluene, isoamyl nitrite, 110 °C, 1.5 h.

reaction mixture. A plausible mechanism of this transformation is represented in Scheme 2.

The structure of the products **4** was confirmed by ¹H, ¹³C, ¹⁹F NMR spectroscopy and mass spectrometry, and by comparison of their spectral data with the previously reported ones for non-fluorinated products.¹³ Ultimately, the structure of the representative compound **4b** was confirmed by X-ray crystallography (Figure 1).[‡]

Compound **4b** is crystallized in trigonal system. Face-to-face $\pi - \pi$ stacking interactions between the pyrido[1,2-*a*]indole fragments are evident, the interplanar separations are in the range of

[†] 2,3-Difluoro-10-(1H-1,2,3-triazol-1-yl)pyrido[1,2-a]indoles **4**. Corresponding triazine **3** (3 mmol) was suspended in dry toluene (130 ml). Isoamyl nitrite (1,61 ml, 12 mmol) was added to this mixture. The mixture was stirred under reflux while solution of 3,4-difluoroanthranilic

acid (2.08 g, 12 mmol) in dry 1,4-dioxane (15 ml) was added dropwise within 30 min. The reaction mixture was refluxed for 1 h, then cooled to room temperature and washed with 3 M KOH solution (3×75 ml), the organic layer was separated and dried with anhydrous Na₂SO₄. The solvents were removed under reduced pressure. Products **4** were isolated by column chromatography [silica gel, AcOEt–CH₂Cl₂ (3:1) as eluent, $R_f = 0.8$]. Analytically pure samples **4** were obtained by the recrystallization from dry acetonitrile. For the analytical data, see Online Supplementary Materials.



3.5 Å, and the glide-related molecules are linked in a head-to-tail fashion to generate a supramolecular architecture of infinite chains (see Figure S1, Online Supplementary Materials). Hydrogen bonds between fluorine and hydrogen atoms are also observed, distances between them are in a range of 2.5–2.6 Å (see Figure S2, Online Supplementary Materials). Similar cases of intermolecular and intramolecular hydrogen bonds involving fluorine atoms were reported.²¹

In conclusion, a convenient one-step approach towards novel 2,3-difluoro-10-(1*H*-1,2,3-triazol-1-yl)pyrido[1,2-*a*]indoles *via* the aryne chemistry has been developed. These compounds can be very promising in view of their biological activities.

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Figure 1 Molecular structure for product 4b.

[‡] Crystal data for **4b**. The single crystal (brown needle, 0.25×0.08×0.02 mm) of compound **4b** (C₂₀H₁₁F₃N₄) was used for X-ray analysis. Analysis was performed at 295(2) K on an Xcalibur E diffractometer using graphite monochromated MoK α (λ = 71.073 pm) and CCD detector. Crystal is trigonal, space group *R*-3 with a = b = 47.097(5) and c = 3.8351(5) Å, $\alpha = \beta =$ = 90°, γ = 120°, V = 7367.1(15) Å³, Z = 18. On the angles 2.29 < θ < 26.36° 10614 reflections were measured, among them 3323 unique reflections ($R_{\text{int}} = 0.0490$), 1679 reflections with $I > 2\sigma(I)$. Completeness to $\theta_{\text{max}} =$ = 26.38 is 99.3%. The structure was solved by direct method and refined by full-matrix least squares at F^2 using the SHELXTL program package. All non-hydrogen atoms were refined anisotropically, the positions of the hydrogen atoms were calculated as a riding model in isotropic approximation. An absorption correction was not applied ($\mu = 0.115 \text{ mm}^{-1}$). Goodness of fit at F^2 1.003; final R values $[I > 2\sigma(I)]$, $R_1 = 0.0548$, $wR_2 = 0.0846$; R value (all reflections) $R_1 = 0.1326$, $wR_2 = 0.1024$. Largest difference peak and hole were 0.143 and $-0.139 \text{ e}\text{\AA}^{-3}$.

CCDC 984078 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre *via* http://www.ccdc.cam.ac.uk.

Online Supplementary Materials

Supplementary data associated with this article can be found in the online version at doi:10.1016/j.mencom.2015.01.003.

References

- (a) R. Ambros, S. Von Angerer and W. Wiegrebe, Arch. Pharm., 1988, 321, 481; (b) R. Ambros, M. R. Schneider and S. Von Angerer, J. Med. Chem., 1990, 33, 153.
- 2 T. W. Hudyma, X. Zheng, F. He, M. Ding, C. P. Bergstrom, P. Hewawasam, S. W. Martin and R. G. Gentles, U.S. Pat. Appl. Publ., 2006, 318, Cont.-In-Part of U.S. Ser. No. 181, 639 (Chem. Abstr., 2006, 145, 189063).
- 3 E. O. M. Orlemans, W. Verboom, M. W. Scheltinga, D. N. Reinhoudt, P. Lelieveld, H. H. Fiebig, B. R. Winterhalter, J. A. Doublell and M. C. Bibbyll, *J. Med. Chem.*, 1989, **32**, 1612.
- 4 (a) S. Donga and W. Wanga, Synth. Met., 2009, **159**, 385; (b) T. V. Trashakhova, E. V. Nosova, M. S. Valova, P. A. Slepukhin, G. N. Lipunova and V. N. Charushin, Russ. J. Org. Chem., 2011, **47**, 753 (Zh. Org. Khim., 2011, **47**, 748).
- 5 G. N. Lipunova, E. V. Nosova, A. A. Laeva and V. N. Charushin, *Pharm. Chem. J.*, 2012, **45**, 709 (*Khim.-Farm. Zh.*, 2012, **45**, 12).
- 6 (a) I. N. Kang, H. K. Shim and T. Zyung, *Chem. Mater.*, 1997, 9, 746; (b) R. M. Gurge, A. M. Sarker, P. M. Lahti, B. Hu and F. E. Karasz, *Macromolecules*, 1997, 30, 8286.
- 7 (a) D. C. Rogness, N. A. Markina, J. P. Waldo and R. C. Larock, J. Org. Chem., 2012, 77, 2743; (b) L.-L. Sun, Z.-Y. Liao, R.-Y. Tang, C.-L. Deng and X.-G. Zhang, J. Org. Chem., 2012, 77, 2850.
- 8 G. T. Sukhanov, G. V. Sakovich, Yu. V. Filippova, I. Yu. Bagryanskaya and A. G. Sukhanova, *Mendeleev Commun.*, 2014, **24**, 280.
- 9 S. G. Agalave, S. R. Maujan and V. S. Pore, *Chemistry An Asian Journal*, 2011, 6, 2696.
- (a) H. Kim, S. Lee, J. Lee and J. Tae, Org. Lett., 2010, 12, 5342;
 (b) K. Ghosh, D. Kar, S. Joardar, D. Sahu and B. Ganquly, RSC Adv., 2013, 3, 16144.
- 11 V. Bhalla, H. Singh and M. Kumar, Dalton Trans., 2012, 41, 11413.
- (a) N. von Braun, *Chem. Ber.*, 1937, **70**, 1760; (b) A. Deegan and F. L. Rose, *J. Chem. Soc. C: Org. Chem.*, 1971, 2756; (c) R. R. Naredla, C. Zheng, S. O. Nilsson Lill and D. A. Klumpp, *J. Am. Chem. Soc.*, 2011, **133**, 13169; (d) A. T. Soldatenkov, M. V. Bagdadi, A. A. Fomichev and N. S. Prostakov, *J. Org. Chem. USSR*, 1982, **18**, 783 (*Zh. Org. Khim.*, 1982, **18**, 902); (e) A. Ohsawa, T. Kawaguchi and H. Igeta, *Chem. Lett.*, 1981, 1737; (f) L. M. Gaster and P. A. Wyman, *Patent US 5998409A*, 1995 (*Chem. Abstr.*, 1995, **122**, 265390).
- 13 I. L. Nikonov, D. S. Kopchuk, I. S. Kovalev, G. V. Zyryanov, A. F. Khasanov, P. A. Slepukhin, V. L. Rusinov and O. N. Chupakhin, *Tetrahedron Lett.*, 2013, **54**, 6427.
- 14 (a) I. S. Kovalev, D. S. Kopchuk, G. V. Zyryanov, P. A. Slepukhin, V. L. Rusinov and O. N. Chupakhin, *Chem. Heterocycl. Compd.*, 2012, 48, 536 (*Khim. Geterotsikl. Soedin.*, 2012, 576); (b) A. V. Dubrovskiy, N. A. Markina and R. C. Larock, *Org. Biomol. Chem.*, 2013, 11, 191.
- (a) K. S. Gebara, G. A. Casagrande and C. Raminelli, *Tetrahedron Lett.*, 2011, **52**, 2849; (b) L. Castedo, S. Escudero, D. Ferez, E. Guitian and D. Pena, *Angew. Chem., Int. Ed. Engl.*, 1997, **37**, 2659; (c) R. C. Larock, Z. Liu and C. Raminelli, *J. Org. Chem.*, 2006, **71**, 4689.
- 16 (a) R. Luo, W. Tang, J. Liao, R. Luo, L. Xie and A. S. Chan, *Chem. Commun.*, 2013, **49**, 9959; (b) H. F. Bettinger, R. P. Bula and I. M. Oppel, *J. Org. Chem.*, 2012, **77**, 3538; (c) R. Webster, C. Boeing and M. Lautens, *J. Am. Chem. Soc.*, 2009, **131**, 444.
- 17 A. A. Layeva, E. V. Nosova, G. N. Lipunova and V. N. Charushin, *Russ. Chem. Bull.*, *Int. Ed.*, 2008, **57**, 947 (*Izv. Akad. Nauk., Ser. Khim.*, 2008, 931).
- 18 (a) U. N. Rao, R. Sathunuru, J. A. Maguire and E. Biehl, *J. Heterocycl. Chem.*, 2004, **41**, 13; (b) D. M. Hodgson, M. W. P. Bebbington and P. Willis, *Org. Biomol. Chem.*, 2003, **1**, 3787; (c) M. W. P. Bebbington, D. M. Hodgson and P. Willis, *Org. Lett.*, 2002, **4**, 4353.
- 19 F. R. Busch, S. J. Hecker, P. R. McGuirk, B. T. O'Neill and H. A. Watson, Jr., Patent EP0342849, 1989 (Chem. Abstr., 1990, 113, 406177).
- 20 (a) F. H. Case, J. Org. Chem., 1965, 30, 931; (b) V. N. Kozhevnikov, O. V. Shabunina, D. S. Kopchuk, M. M. Ustinova, B. König and D. N. Kozhevnikov, *Tetrahedron*, 2008, 64, 8963; (c) D. S. Kopchuk, G. V. Zyryanov, I. S. Kovalev, A. F. Khasanov, A. S. Medvedevskikh, V. L. Rusinov and O. N. Chupakhin, *Chem. Heterocycl. Compd.*, 2013, 49, 500 (*Khim. Geterotsikl. Soedin.*, 2013, 535).
- 21 (a) C. Dalvit and A. Vulpetti, *ChemMedChem*, 2012, 7, 262; (b) J. D. Dunitz and R. Taylor, *Chem. Eur. J.*, 1997, 3, 89.

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